Standard Operating Procedures

Small Sputtering Machine

Ross Levine

With revisions by: Brian Wajdyk

Important

- Gloves should be worn while handling anything going in vacuum to reduce contamination.
- You can only use CeNSE laboratories and equipment if you have been approved by Brian, reserved the tool on the calendar, and filled out a reservation form. No Exceptions!
- If the equipment is acting unusual STOP! Place the instrument in the normal standby mode. Leave a note on the machine. Then please discuss with Brian before proceeding.
- Any accidental damage must be reported immediately.
- All CeNSE laboratories are protected by video surveillance.



Machine Exterior

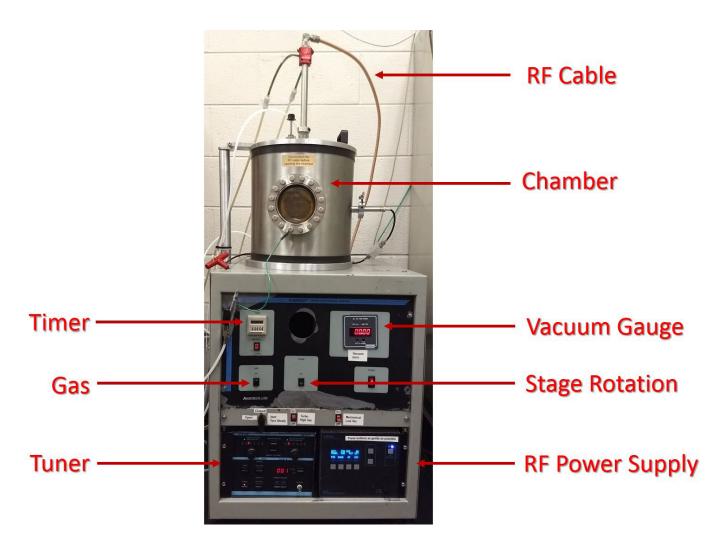


Figure 1 Machine exterior



Figure 1a Power and tuning controls



Figure 1b Gas control



Figure 1c Vent and Pump panel



Figure 1d Turbo Pump controller

Other Parts of the Machine



Figure 2a Top of chamber



Figure 2b Inside of Chamber



Figure 2c Thickness Monitor

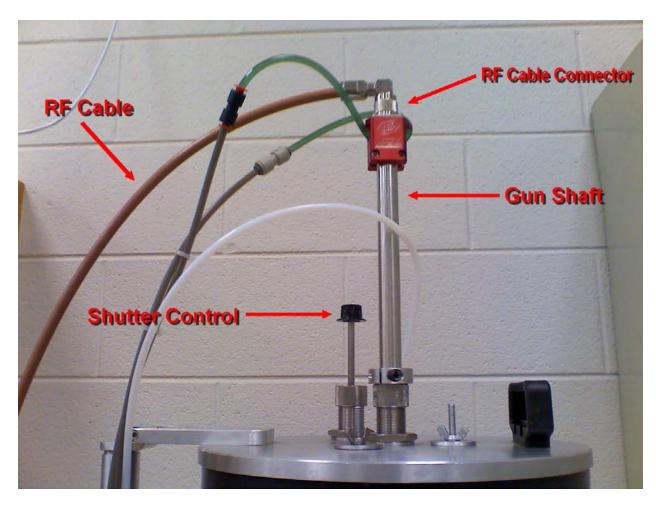


Figure 2d Chamber lid

Operating Procedure

1. Preparation

1.1. Before beginning, be sure to wear gloves to reduce contamination and to protect the hands.

2. Load the sample and target

- 2.1. Venting the System:
 - 2.1.1. Disable the turbo pump by flipping the switch downward on the main panel (figure 1c).
 Wait about a minute before proceeding.
 - 2.1.2. Open the vent valve (figure 1c)
 - 2.1.3. Disable the rough pump by flipping the switch downward on the main panel (figure 1c).
 - 2.1.4. Look through the chamber window and ensure that the shutter is directly below the gun; if it is not, the shutter will collide with the chamber wall when opening the lid.
 - 2.1.5. Wait until the chamber is vented. There will be no obvious signs, but the chamber lid will not open if the chamber is not fully vented. Venting time is typically between 2-5 minutes.
 - 2.1.6. Turn off the venting gas (Figure 1c)
 - 2.1.7. Using a step ladder, unscrew the RF cable (Figure 2d). Place the cable out of the way.
 - 2.1.8. Grab the chamber lid handle and open the chamber lid completely; it will sit upsidedown on its own pole.

2.2. Load the sample and target

- 2.2.1. Place the sample on the stage inside the chamber (figure 2b). Ensure that the side on which the deposition will take place is facing upwards.
- 2.2.2. Before placing your target on the gun, check to make sure there is thermal paste on the copper cathode. The proper amount is thin enough to cover it completely but, it should not be completely opaque. You should still be able to see the copper color through the gray paste.
- 2.2.3. Place your target on the gun, with the magnetic keeper facing the gun (figure 2a).
- 2.3. Pumping down the chamber:
 - 2.3.1. Close the chamber lid using the handle. The lid is centered over the chamber the lid sticks out all the way around the chamber rim.
 - 2.3.2. Replace the RF cable (Figure 2d).
 - 2.3.3. Turn on the rough pump by flipping the switch upward on the main panel (figure 1c).
 - 2.3.4. Wait until the vacuum gauge turns on. The blinking of "0000" will stop and numbers will be present. Turn on the turbo pump by flipping the switch upward (figure 1c).

3. Configure the deposition (figure 2c)

- 3.1. Set the deposition thickness:
 - 3.1.1. Press and hold the thickness button on the deposition monitor.
 - 3.1.2. Use the increase and decrease buttons to set the desired thickness.

- 3.2. Set the material density:
 - 3.2.1. Use the chart in Appendix A to find the density of the deposition material.
 - 3.2.2. Press and hold the density button on the deposition monitor.
 - 3.2.3. Use the increase and decrease buttons to set the density (figure 2c).
- 3.3. Set the material's acoustic impedance:
 - 3.3.1. Use the chart in Appendix A to find the acoustic impedance of the deposition material.
 - 3.3.2. Press and hold the acoustic impedance button on the deposition monitor (figure 2c).
 - 3.3.3. Use the increase and decrease buttons to set the density (figure 2c).
- 3.4. Set the tooling factor:
 - 3.4.1. Tooling factor must be found experimentally and varies by material. See Appendix B
 - 3.4.2. Press and hold the tooling factor % button on the deposition monitor (figure 2c).
 - 3.4.3. Use the increase and decrease buttons to set the tooling factor (figure 2c).
- 3.5. Press the start button on the deposition monitor. Note shutter open light should be on.

4. Set the RF output power (figure 1a)

4.1. Press the up arrow to increase or down arrow to decrease (figure 1a) the RF power wattage. If you aren't sure, 75watts is a good place to start.

5. Set the gas flow (figure 1b)

- 5.1. Edit the gas flow if necessary. 30sccm is default. Status of on is default. If no change is necessary, skip to step 6
 - 5.1.1. Press the $\leftarrow \rightarrow$ buttons on the gas controller to highlight the setpoint of Channel 5.
 - 5.1.2. Use the $\downarrow \uparrow$ buttons to adjust gas flow.
 - 5.1.3. Press the $\leftarrow \rightarrow$ buttons on the gas controller to highlight the status of Channel 5.
 - 5.1.4. Use the $\downarrow \uparrow$ buttons to change to on.

6. Start the sputtering process

- 6.1. Ensure that the valve is open on the argon tank. Open it by turning the grey knob at the top clockwise.
- 6.2. Turn on the "gas" switch. (figure 1)
- 6.3. Wait for the vacuum to stabilize inside the chamber.
- 6.4. Turn on the "stage" switch for an isotropic effect. This will to allow the stage to rotate creating better step coverage and be more conformal to the sample topography (default). When this isn't desired, such as in a lift-off process, this can be left off. (figure 1)
- 6.5. Check the RF source display for the letters "EXT" are not present to ensure the interlocks are open (Figure 1a).
- 6.6. Press the "RF" button (Figure 1a) to start the plasma.
- 6.7. Pre-sputter with the shutter closed to remove surface contaminants from the target. 1-2 minutes is usually a good amount of time. The vacuum level should be stable.
- 6.8. When you are ready to begin, turn the shutter fully clockwise open (figure 2d), and quickly press start again on the deposition monitor to re-zero the measurement. The thickness monitor will automatically turn off the RF power when the desired thickness is reached.

7. End the sputtering process (Figure 1)

- 7.1. Turn off components, in this order:
- 7.2. Turn off the gas:
 - 7.2.1. Turn off the "gas" switch.
 - 7.2.2. Turn off the gas flow (figure 1b)
 - 7.2.3. Close the valve on the argon tank.
- 7.3. Turn off the "stage" switch
- 7.4. Turn off the turbo pump and wait about a minute before proceeding to the next step.
- 7.5. Slowly open the venting gas by rotating to the horizontal position.
- 7.6. Turn off the rough pump.
- 7.7. Move the shutter back over the gun.
- 7.8. Wait for the chamber to pressurize.
- 7.9. Rotate the vent knob to the vertical position. This will turn off the venting gas.
- 7.10. Open the chamber lid. Look through the chamber window and ensure that the shutter is directly below the gun; if it is not, the shutter will collide with the chamber wall when opening the lid.
- 7.11. Unload the sample if you are done with it (using gloves).
- 7.12. If it is no longer needed, remove the target and put it back in its container. If necessary, carefully use the tool that is attached to the machine to pry it out. Be careful to not scratch the gun.
- 7.13. Close the chamber lid using the handle. The lid is centered over the chamber the lid sticks out all the way around the chamber rim.
- 7.14. Pump down the chamber as in step 2.3

Appendix A: Material Properties

Use the chart below to find the properties of the deposition material for step 3 in the Operating Procedure.

| | | | Acoustic | |
|---------|---------|---------|-----------|-------------------------|
| Formula | Density | Z-Ratio | Impedance | Material Name |
| Ag | 10.5 | 0.529 | 16.69 | Silver |
| AgBr | 6.47 | 1.18 | 7.48 | Silver Bromide |
| AgCl | 5.56 | 1.32 | 6.69 | Silver Chloride |
| Al | 2.70 | 1.08 | 8.18 | Aluminum |
| Al2O3 | 3.97 | 0.336 | 26.28 | Aluminum Oxide |
| Al4C3 | 2.36 | ? | | Aluminum Carbide |
| AlF3 | 3.07 | ? | | Aluminum Fluoride |
| AIN | 3.26 | ? | | Aluminum Nitride |
| AlSb | 4.36 | 0.743 | 11.88 | Aluminum Antimonide |
| As | 5.73 | 0.966 | 9.14 | Arsenic |
| As2Se3 | 4.75 | ? | | Arsenic Selenide |
| Au | 19.3 | 0.381 | 23.18 | Gold |
| В | 2.37 | 0.389 | 22.70 | Boron |
| B2O3 | 1.82 | ? | | Boron Oxide |
| B4C | 2.37 | ? | | Boron Carbide |
| BN | 1.86 | ? | | Boron Nitride |
| Ва | 3.5 | 2.1 | 4.20 | Barium |
| BaF2 | 4.886 | 0.793 | 11.13 | Barium Fluoride |
| BaN2O6 | 3.244 | 1.261 | 7.00 | Barium Nitrate |
| BaO | 5.72 | ? | | Barium Oxide |
| BaTiO3 | 5.999 | 0.464 | 19.03 | Barium Titanate (Tetr) |
| BaTiO3 | 6.035 | 0.412 | 21.43 | Barium Titanate (Cubic) |
| Ве | 1.85 | 0.543 | 16.26 | Beryllium |
| BeF2 | 1.99 | ? | | Beryllium Fluoride |
| BeO | 3.01 | ? | | Beryllium Oxide |
| Bi | 9.8 | 0.79 | 11.18 | Bismuth |
| Bi2O3 | 8.9 | ? | | Bismuth Oxide |
| Bi2S3 | 7.39 | ? | | Bismuth Trisuiphide |
| Bi2Se3 | 6.82 | ? | | Bismuth Selenide |

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 $^{^{\}rm 1}$ SQM-160 User's Guide, Version 4.06. Sigma Instruments, Inc. 2000-2008.

| | | | Acoustic | |
|----------|---------|---------|-----------|----------------------------|
| Formula | Density | Z-Ratio | Impedance | Material Name |
| Bi2Te3 | 7.7 | ? | | Bismuth Telluride |
| BiF3 | 5.32 | ? | | Bismuth Fluoride |
| С | 2.25 | 3.26 | 2.71 | Carbon (Graphite) |
| С | 3.52 | 0.22 | 40.14 | Carbon (Diamond) |
| C8H8 | 1.1 | ? | | Parlyene (Union Carbide) |
| Ca | 1.55 | 2.62 | 3.37 | Calcium |
| CaF2 | 3.18 | 0.775 | 11.39 | Calcium Fluoride |
| CaO | 3.35 | ; | | Calcium Oxide |
| CaO-SiO2 | 2.9 | , | | Calcium Silicate (3) |
| CaSO4 | 2.962 | 0.955 | 9.25 | Calcium Sulfate |
| CaTiO3 | 4.1 | , | | Calcium Titanate |
| CaWO4 | 6.06 | ? | | Calcium Tungstate |
| Cd | 8.64 | 0.682 | 12.95 | Cadmium |
| CdF2 | 6.64 | ? | | Cadmium Fluoride |
| CdO | 8.15 | ? | | Cadmium Oxide |
| CdS | 4.83 | 1.02 | 8.66 | Cadmium Sulfide |
| CdSe | 5.81 | ? | | Cadmium Selenide, |
| CdTe | 6.2 | 0.98 | 9.01 | Cadmium Telluride |
| Ce | 6.78 | ? | | Cerium |
| CeF3 | 6.16 | ? | | Cerium (III) Fluoride |
| CeO2 | 7.13 | ? | | Cerium (IV) Dioxide |
| Со | 8.9 | 0.343 | 25.74 | Cobalt |
| CoO | 6.44 | 0.412 | 21.43 | Cobalt Oxide |
| Cr | 7.2 | 0.305 | 28.95 | Chromium |
| Cr2O3 | 5.21 | ? | | Chromium (III) Oxide |
| Cr3C2 | 6.68 | ? | | Chromium Carbide |
| CrB | 6.17 | ? | | Chromium Boride |
| Cs | 1.87 | ? | | Cesium |
| Cs2SO4 | 4.243 | 1.212 | 7.29 | Cesium Sulfate |
| CsBr | 4.456 | 1.41 | 6.26 | Cesium Bromide |
| CsCl | 3.988 | 1.399 | 6.31 | Cesium Chloride |
| CsI | 4.516 | 1.542 | 5.73 | Cesium Iodide |
| Cu | 8.93 | 0.437 | 20.21 | Copper |
| Cu2O | 6 | ? | | Copper Oxide |
| Cu2S | Cu2S | 5.6 | 1.58 | Copper (I) Sulfide (Alpha) |
| Cu2S | Cu2S | 5.8 | 1.52 | Copper (I) Sulfide (Beta) |
| CuS | CuS | 4.6 | 1.92 | Copper (II) Sulfide |
| Dy | Dy | 8.55 | 1.03 | Dysprosium |
| Dy2O3 | Dy2O3 | 7.81 | 1.13 | Dysprosium Oxide |
| Er | Er | 9.05 | 0.98 | Erbium |

| | | | Acoustic | |
|---------|---------|---------|-----------|---------------------|
| Formula | Density | Z-Ratio | Impedance | Material Name |
| Er2O3 | Er2O3 | 8.64 | 1.02 | Erbium Oxide |
| Eu | Eu | 5.26 | 1.68 | Europium |
| EuF2 | EuF2 | 6.5 | 1.36 | Europium Fluoride |
| Fe | 7.86 | 0.349 | 25.30 | Iron |
| Fe2O3 | 5.24 | ? | | Iron Oxide |
| FeO | 5.7 | ? | | Iron Oxide |
| FeS | 4.84 | ? | | Iron Sulphide |
| Ga | 5.93 | 0.593 | 14.89 | Gallium |
| Ga2O3 | 5.88 | 3 | | Gallium Oxide (B) |
| GaAs | 5.31 | 1.59 | 5.55 | Gallium Arsenide |
| GaN | 6.1 | ? | | Gallium Nitride |
| GaP | 4.1 | ? | | Gallium Phosphide |
| GaSb | 5.6 | ? | | Gallium Antimonide |
| Gd | 7.89 | 0.67 | 13.18 | Gadolinium |
| Gd2O3 | 7.41 | ? | | Gadolinium Oxide |
| Ge | 5.35 | 0.516 | 17.11 | Germanium |
| Ge3N2 | 5.2 | ? | | Germanium Nitride |
| GeO2 | 6.24 | ? | | Germanium Oxide |
| GeTe | 6.2 | ? | | Germanium Telluride |
| Hf | 13.09 | 0.36 | 24.53 | Hafnium |
| HfB2 | 10.5 | ? | | Hafnium Boride, |
| HfC | 12.2 | ? | | Hafnium Carbide |
| HfN | 13.8 | ? | | Hafnium Nitride |
| HfO2 | 9.68 | ? | | Hafnium Oxide |
| HfSi2 | 7.2 | 3 | | Hafnium Silicide |
| Hg | 13.46 | 0.74 | 11.93 | Mercury |
| Но | 8.8 | 0.58 | 15.22 | Holminum |
| Ho2O3 | 8.41 | ? | | Holminum Oxide |
| In | 7.3 | 0.841 | 10.50 | Indium |
| In2O3 | 7.18 | ? | | Indium Sesquioxide |
| In2Se3 | 5.7 | ? | | Indium Selenide |
| In2Te3 | 5.8 | ? | | Indium Telluride |
| InAs | 5.7 | ? | | Indium Arsenide |
| InP | 4.8 | ? | | Indium Phosphide |
| InSb | 5.76 | 0.769 | 11.48 | Indium Antimonide |
| Ir | 22.4 | 0.129 | 68.45 | Iridium |
| K | 0.86 | 10.189 | 0.87 | Potassium |
| KBr | 2.75 | 1.893 | 4.66 | Potassium Bromide |
| KCl | 1.98 | 2.05 | 4.31 | Potassium Chloride |
| KF | 2.48 | ? | | Potassium Fluoride |

| Nation | | | | Acoustic | |
|---|-----------|-------|--|----------|------------------------|
| La 6.17 0.92 9.60 Lanthanum La2O3 6.51 ? Lanthanum Oxide LaB6 2.61 ? Lanthanum Boride LaF3 5.94 ? Lanthanum Fluoride Li 0.53 5.9 1.50 Lithium Fluoride LiBr 3.47 1.23 7.18 Lithium Fluoride LiF 2.638 0.778 11.35 Lithium Fluoride LiF 2.638 0.778 11.35 Lithium Fluoride LiNDO3 4.7 0.463 19.07 Lithium Niobate Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgAl2O4 3.6 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Aluminate MgAl2O6 8 ? Spinel MgP2 3.18 0.637 13.86 Magnesium Oxide Mn 7.2 0.377 23.42 Mangan | Formula | • | | | |
| La2O3 6.51 ? Lanthanum Oxide LaB6 2.61 ? Lanthanum Boride LaF3 5.94 ? Lanthanum Fluoride Li 0.53 5.9 1.50 Lithium LiBr 3.47 1.23 7.18 Lithium Bromide LiF 2.638 0.778 11.35 Lithium Fluoride LiNbO3 4.7 0.463 19.07 Lithium Fluoride Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium Sluminate MgAl2O4 3.6 ? Magnesium Magnesium Sluminate MgAl2O6 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Sluminate MnO 7.2 0.377 23.42 Manganese (Discussion Suprate Mn 7.2 0.377 23.42 Manganese (Discussion Suprate MnO 5.39 | KI | 3.128 | 2.077 | 4.25 | Potassium Iodide |
| LaB6 2.61 ? Lanthanum Boride LaF3 5.94 ? Lanthanum Fluoride Li 0.53 5.9 1.50 Lithium LiBr 3.47 1.23 7.18 Lithium Bromide LiF 2.638 0.778 11.35 Lithium Bromide LiF 2.638 0.778 11.35 Lithium Bromide LiNbO3 4.7 0.463 19.07 Lithium Bromide Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgP12 3.6 ? Magnesium Aluminate MgA1204 3.6 ? Magnesium Aluminate Mg206 8 ? Spinel Mg20 3.58 0.411 21.48 Magnesium Fluoride Mn 7.2 0.377 23.42 Manganese Oxide Mn 7.2 0.377 23.42 Manganese Oxide Mn 7.2 0.467 18.91 M | | | 1 | 9.60 | |
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| Li 0.53 5.9 1.50 Lithium LiBr 3.47 1.23 7.18 Lithium Bromide LiF 2.638 0.778 11.35 Lithium Fluoride LiNbO3 4.7 0.463 19.07 Lithium Niobate Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgAl2O4 3.6 ? Magnesium Aluminate MgAl2O6 8 ? Spinel MgP2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Garbide Mo2C 9.18 ? Molybdenum Garbide Mo2Z 4.8 | LaB6 | 2.61 | ? | | Lanthanum Boride |
| LiBr 3.47 1.23 7.18 Lithium Bromide LiF 2.638 0.778 11.35 Lithium Fluoride LiNDO3 4.7 0.463 19.07 Lithium Niobate Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgA1204 3.6 ? Magnesium Aluminate MgA1206 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese (II) Sulfide Mn 7.2 0.377 23.42 Manganese (II) Sulfide Mn 5.39 0.467 18.91 Manganese (II) Sulfide Mn 10.2 0.257 34.36 Molybdenum Earbide Mo2C 9.18 ? Molybdenum Earbide Mo2A | LaF3 | 5.94 | 3 | | Lanthanum Fluoride |
| LiF 2.638 0.778 11.35 Lithium Fluoride LiNbO3 4.7 0.463 19.07 Lithium Niobate Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgAJ204 3.6 ? Magnesium Aluminate MgAJ206 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide Mo82 7.12 ? Molybdenum Trioxidie MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 | Li | 0.53 | 5.9 | 1.50 | Lithium |
| LiNbO3 4.7 0.463 19.07 Lithium Niobate Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgAI2O4 3.6 ? Magnesium Aluminate MgAI2O6 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Fluoride Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese Cvide MnS 3.99 0.94 9.39 Molybdenum Ecride Mo2C | LiBr | 3.47 | 1.23 | 7.18 | Lithium Bromide |
| Lu 9.84 ? Lutetium Mg 1.74 1.61 5.48 Magnesium MgAl2O4 3.6 ? Magnesium Aluminate MgAl2O6 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide Mo2C 9.18 ? Molybdenum Carbide Mo82 7.12 ? Molybdenum Toxidide MoS2 4.8 ? Molybdenum Toxidide NaSALF14 2.9 ? Cryolite NaSAL3F14 2.9 ? Chiolite NaCI <td< td=""><td>LiF</td><td>2.638</td><td>0.778</td><td>11.35</td><td>Lithium Fluoride</td></td<> | LiF | 2.638 | 0.778 | 11.35 | Lithium Fluoride |
| Mg 1.74 1.61 5.48 Magnesium MgAl2O4 3.6 ? Magnesium Aluminate MgAl2O6 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese Oxide MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide Mo82 7.12 ? Molybdenum Boride Mo3 4.7 ? Molybdenum Trioxidide Mo52 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium NaSAIF14 2.9 ? Cryolite NaSAIF14 2.9 ? Cryolite | LiNbO3 | 4.7 | 0.463 | 19.07 | Lithium Niobate |
| MgAl2O4 3.6 ? Magnesium Aluminate MgAl2O6 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese (II) Sulfide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride MoS2 4.8 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Trioxdide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite NaBr 3.2 ? Chiolite NaClO3 2.164 1.565 5.62 Sodium Chloride <tr< td=""><td>Lu</td><td>9.84</td><td>?</td><td></td><td>Lutetium</td></tr<> | Lu | 9.84 | ? | | Lutetium |
| MgAl2O6 8 ? Spinel MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide Mo82 7.12 ? Molybdenum Boride Mo03 4.7 ? Molybdenum Trioxidie Mo3 4.7 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium NaSALF6 2.9 ? Cryolite NaSAL3F14 2.9 ? Cryolite NaCI 2.17 1.57 5.62 Sodium Bromide NaCI 2.164 1.565 5.64 Sodium Chlorate | Mg | 1.74 | 1.61 | 5.48 | Magnesium |
| MgF2 3.18 0.637 13.86 Magnesium Fluoride MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride Mo03 4.7 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AIF6 2.9 ? Cryolite NaBr 3.2 ? Cryolite NaCl 2.17 1.57 5.62 Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chlorate NaF 2.558 0.949 9.30 | MgAl2O4 | 3.6 | 3 | | Magnesium Aluminate |
| MgO 3.58 0.411 21.48 Magnesium Oxide Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride MoD3 4.7 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite NaBr 3.2 ? Cryolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaCl 2.164 1.565 5.64 Sodium Chlorate NaP 2.558 0.949 9.30 Sodium Fluoride | MgAl2O6 | 8 | ? | | Spinel |
| Mn 7.2 0.377 23.42 Manganese MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride Mo03 4.7 ? Molybdenum Disulfide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite NaSAL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaCl 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate | MgF2 | 3.18 | 0.637 | 13.86 | Magnesium Fluoride |
| MnO 5.39 0.467 18.91 Manganese Oxide MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride MoO3 4.7 ? Molybdenum Disulfide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (C | MgO | 3.58 | 0.411 | 21.48 | Magnesium Oxide |
| MnS 3.99 0.94 9.39 Manganese (II) Sulfide Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride MoO3 4.7 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na O.97 4.8 1.84 Sodium Na O.97 4.8 1.84 Sodium Na O.97 4.8 1.84 Sodium Na Sodium Disulfide Cryolite Cryolite Na Sodium Disulfide Cryolite Cryolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaCl 2.164 1.565 5.64 Sodium Chlorate NaNO3 2.27 1.194 7.40 Sodium Ritrate Nb 8.578 0.492 | Mn | 7.2 | 0.377 | 23.42 | Manganese |
| Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride Mo03 4.7 ? Molybdenum Trioxdide Mo52 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Columbium NbC <td>MnO</td> <td>5.39</td> <td>0.467</td> <td>18.91</td> <td>Manganese Oxide</td> | MnO | 5.39 | 0.467 | 18.91 | Manganese Oxide |
| Mo 10.2 0.257 34.36 Molybdenum Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride Mo03 4.7 ? Molybdenum Trioxdide Mo52 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Columbium NbC <td>MnS</td> <td>3.99</td> <td>0.94</td> <td>9.39</td> <td>Manganese (II) Sulfide</td> | MnS | 3.99 | 0.94 | 9.39 | Manganese (II) Sulfide |
| Mo2C 9.18 ? Molybdenum Carbide MoB2 7.12 ? Molybdenum Boride MoO3 4.7 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaCl03 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (V) Oxide Nb2O5 4.47 ? Niobium (V) Oxide NbC 7.82 ? Niobium Carbide NbN 8.4 <td>Мо</td> <td>10.2</td> <td>0.257</td> <td>34.36</td> <td></td> | Мо | 10.2 | 0.257 | 34.36 | |
| MoB2 7.12 ? Molybdenum Boride MoO3 4.7 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaCl03 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (Columbium) Nb2O5 4.47 ? Niobium (V) Oxide NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 | Mo2C | 9.18 | ? | | Molybdenum Carbide |
| MoO3 4.7 ? Molybdenum Trioxdide MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (Columbium) Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Carbide NbN 8.4 ? Niobium Nitride NbN 8.4 ? Niobium Nitride NbO 7.24 | MoB2 | 7.12 | ? | | · |
| MoS2 4.8 ? Molybdenum Disulfide Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium Wirride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride NbN 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | MoO3 | 4.7 | ? | | • |
| Na 0.97 4.8 1.84 Sodium Na3AlF6 2.9 ? Cryolite Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (Columbium) Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Carbide NbC 7.82 ? Niobium Nitride Nb 8.4 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | MoS2 | 4.8 | ? | | · |
| Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (Columbium) Nb2O3 7.5 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | Na | 0.97 | 4.8 | 1.84 | - |
| Na5AL3F14 2.9 ? Chiolite NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (Columbium) Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | Na3AlF6 | 2.9 | ? | | Cryolite |
| NaBr 3.2 ? Sodium Bromide NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium (Columbium) Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | Na5AL3F14 | 2.9 | ? | | |
| NaCl 2.17 1.57 5.62 Sodium Chloride NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | NaBr | | | | |
| NaClO3 2.164 1.565 5.64 Sodium Chlorate NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | | 5.62 | |
| NaF 2.558 0.949 9.30 Sodium Fluoride NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | | | |
| NaNO3 2.27 1.194 7.40 Sodium Nitrate Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | | | |
| Nb 8.578 0.492 17.95 Niobium (Columbium) Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | | | |
| Nb2O3 7.5 ? Niobium Trioxide Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | | | |
| Nb2O5 4.47 ? Niobium (V) Oxide NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | | | , , |
| NbB2 6.97 ? Niobium Boride NbC 7.82 ? Niobium Carbide NbN 8.4 ? Niobium Nitride Nd 7 ? Neodynium Nd2O3 7.24 ? Neodynium Oxide | | | + | | |
| NbC7.82?Niobium CarbideNbN8.4?Niobium NitrideNd7?NeodyniumNd2O37.24?Neodynium Oxide | | | - | | . , |
| NbN8.4?Niobium NitrideNd7?NeodyniumNd2O37.24?Neodynium Oxide | | | + | | |
| Nd7?NeodyniumNd2O37.24?Neodynium Oxide | | | + | | |
| Nd2O3 7.24 ? Neodynium Oxide | | | + | | |
| | | | | | · |
| NUCS I DOUD I Y I INFORMATION HITOYIGA | NdF3 | 6.506 | ? | | Neodynium Fluoride |

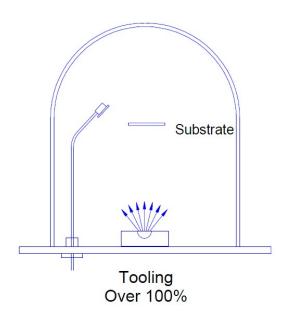
| | | | Acoustic | |
|---------|---------|---------|-----------|---------------------|
| Formula | Density | Z-Ratio | Impedance | Material Name |
| Ni | 8.91 | 0.331 | 26.68 | Nickel |
| NiCr | 8.5 | ? | | Nichrome |
| NiCrFe | 8.5 | ? | | Inconel |
| NiFe | 8.7 | ? | | Permalloy |
| NiFeMo | 8.9 | ? | | Supermalloy |
| NiO | 7.45 | ? | | Nickel Oxide |
| P3N5 | 2.51 | ? | | Phosphorus Nitride |
| Pb | 11.3 | 1.13 | 7.81 | Lead |
| PbCl2 | 5.85 | ? | | Lead Chloride |
| PbF2 | 8.24 | 0.661 | 13.36 | Lead Fluoride |
| PbO | 9.53 | ? | | Lead Oxide |
| PbS | 7.5 | 0.566 | 15.60 | Lead Sulfide |
| PbSe | 8.1 | ? | | Lead Selenide |
| PbSnO3 | 8.1 | ? | | Lead Stannate |
| PbTe | 8.16 | 0.651 | 13.56 | Lead Telluride |
| Pd | 12.038 | 0.357 | 24.73 | Palladium |
| PdO | 8.31 | ? | | Palladium Oxide |
| Ро | 9.4 | ? | | Polonium |
| Pr | 6.78 | ? | | Praseodymium |
| Pr2O3 | 6.88 | ? | | Praseodymium Oxide |
| Pt | 21.4 | 0.245 | 36.04 | Platinum |
| PtO2 | 10.2 | ? | | Platinum Oxide |
| Ra | 5 | ? | | Radium |
| Rb | 1.53 | 2.54 | 3.48 | Rubidium |
| Rbl | 3.55 | ? | | Rubidium Iodide |
| Re | 21.04 | 0.15 | 58.87 | Rhenium |
| Rh | 12.41 | 0.21 | 42.05 | Rhodium |
| Ru | 12.362 | 0.182 | 48.52 | Ruthenium |
| S8 | 2.07 | 2.29 | 3.86 | Sulphur |
| Sb | 6.62 | 0.768 | 11.50 | Antimony |
| Sb2O3 | 5.2 | ? | | Antimony Trioxide |
| Sb2S3 | 4.64 | ? | | Antimony Trisulfide |
| Sc | 3 | 0.91 | 9.70 | Scandium |
| Sc2O3 | 3.86 | ? | | Scandium Oxide |
| Se | 4.81 | 0.864 | 10.22 | Selenium |
| Si | 2.32 | 0.712 | 12.40 | Silicon |
| Si3N4 | 3.44 | *1000 | | Silicon Nitride |
| SiC | 3.22 | ? | | Silicon Carbide |
| SiO | 2.13 | 0.87 | 10.15 | Silicon (II) Oxide |
| SiO2 | 2.648 | 1 | 8.83 | Silicon Dioxide |

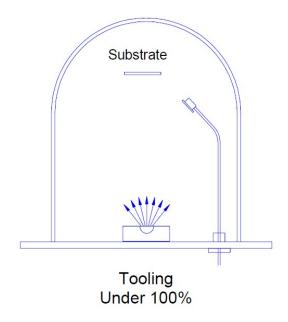
| | | | Acoustic | |
|---------|---------|---------|-----------|-----------------------|
| Formula | Density | Z-Ratio | Impedance | Material Name |
| Sm | 7.54 | 0.89 | 9.92 | Samarium |
| Sm2O3 | 7.43 | ? | | Samarium Oxide |
| Sn | 7.3 | 0.724 | 12.20 | Tin |
| SnO2 | 6.95 | ? | | Tin Oxide |
| SnS | 5.08 | , | | Tin Sulfide |
| SnSe | 6.18 | , | | Tin Selenide |
| SnTe | 6.44 | ? | | Tin Telluride |
| Sr | 2.6 | ? | | Strontium |
| SrF2 | 4.277 | 0.727 | 12.15 | Strontium Fluroide |
| SrO | 4.99 | 0.517 | 17.08 | Strontium Oxide |
| Та | 16.6 | 0.262 | 33.70 | Tantalum |
| Ta2O5 | 8.2 | 0.3 | 29.43 | Tantalum (V) Oxide |
| TaB2 | 11.15 | , | | Tantalum Boride |
| TaC | 13.9 | , | | Tantalum Carbide |
| TaN | 16.3 | , | | Tantalum Nitride |
| Tb | 8.27 | 0.66 | 13.38 | Terbium |
| Tc | 11.5 | ? | | Technetium |
| Te | 6.25 | 0.9 | 9.81 | Tellurium |
| TeO2 | 5.99 | 0.862 | 10.24 | Tellurium Oxide |
| Th | 11.694 | 0.484 | 18.24 | Thorium |
| ThF4 | 6.32 | ? | | Thorium (IV) Fluoride |
| ThO2 | 9.86 | 0.284 | 31.09 | Thorium Dioxide |
| ThOF2 | 9.1 | ? | | Thorium Oxyfluoride |
| Ti | 4.5 | 0.628 | 14.06 | Titanium |
| Ti2O3 | 4.6 | ? | | Titanium Sesquioxide |
| TiB2 | 4.5 | ? | | Titanium Boride |
| TiC | 4.93 | ? | | Titanium Carbide |
| TiN | 5.43 | ? | | Titanium Nitride |
| TiO | 4.9 | ? | | Titanium Oxide |
| TiO2 | 4.26 | 0.4 | 22.08 | Titanium (IV) Oxide |
| TI | 11.85 | 1.55 | 5.70 | Thallium |
| TlBr | 7.56 | ? | | Thallium Bromide |
| TICI | 7 | ? | | Thallium Chloride |
| TII | 7.09 | ? | | Thallium Iodide (B) |
| U | 19.05 | 0.238 | 37.10 | Uranium |
| U3O8 | 8.3 | ? | | Tri Uranium Octoxide |
| U4O9 | 10.969 | 0.348 | 25.37 | Uranium Oxide |
| UO2 | 10.97 | 0.286 | 30.87 | Uranium Dioxide |
| V | 5.96 | 0.53 | 16.66 | Vanadium |
| V2O5 | 3.36 | ? | | Vanadium Pentoxide |

| | | | Acoustic | |
|---------|---------|---------|-----------|---------------------|
| Formula | Density | Z-Ratio | Impedance | Material Name |
| VB2 | 5.1 | ? | | Vanadium Boride |
| VC | 5.77 | ? | | Vanadium Carbide |
| VN | 6.13 | ? | | Vanadium Nitride |
| VO2 | 4.34 | ? | | Vanadium Dioxide |
| W | 19.3 | 0.163 | 54.17 | Tungsten |
| WB2 | 10.77 | ? | | Tungsten Boride |
| WC | 15.6 | 0.151 | 58.48 | Tungsten Carbide |
| WO3 | 7.16 | ? | | Tungsten Trioxide |
| WS2 | 7.5 | ? | | Tungsten Disulphide |
| WSi2 | 9.4 | ? | | Tungsten Suicide |
| Υ | 4.34 | 0.835 | 10.57 | Yttrium |
| Y2O3 | 5.01 | ? | | Yttrium Oxide |
| Yb | 6.98 | 1.13 | 7.81 | Ytterbium |
| Yb2O3 | 9.17 | ? | | Ytterbium Oxide |
| Zn | 7.04 | 0.514 | 17.18 | Zinc |
| Zn3Sb2 | 6.3 | ? | | Zinc Antimonide |
| ZnF2 | 4.95 | ? | | Zinc Fluoride |
| ZnO | 5.61 | 0.556 | 15.88 | Zinc Oxide |
| ZnS | 4.09 | 0.775 | 11.39 | Zinc Sulfide |
| ZnSe | 5.26 | 0.722 | 12.23 | Zinc Selenide |
| ZnTe | 6.34 | 0.77 | 11.47 | Zinc Telluride |
| Zr | 6.49 | 0.6 | 14.72 | Zirconium |
| ZrB2 | 6.08 | ? | | Zirconium Boride |
| ZrC | 6.73 | 0.264 | 33.45 | Zirconium Carbide |
| ZrN | 7.09 | ? | | Zirconium Nitride |
| ZrO2 | 5.6 | ? | | Zirconium Oxide |

Appendix B: Tooling Factor

What is Tooling Factor? Tooling Factor is a correction for the difference in material deposited on the quartz sensor versus the substrate. Illustrated below is an example of how difference in distance between the sensor and substrate causes an incorrect reading as you would see in an electron or thermal evaporation system. It is impossible to place a sensor in exactly the same place as your substrate unless the sensor is your substrate.





How do I determine Tooling Factor?

- 1. Place your substrate and a sensor in their normal position. Mask part of the substrate with a thin material. Thinner is better (i.e. microscope cover glass).
- 2. Set Tooling to an approximate value or if unknown use 100. ($Tooling_{Approximate}$)
- Set Density and Z-Factor for your material. Deposit 1000Å or more of material. $(Thickness_{QCM})$
- Use a profilometer or AFM to measure the substrate's actual film thickness. (*Thickness*_{Actual})
- 6. The Tooling Factor is calculated by:

 $Thickness_{Actual}$

 $Tooling_{Actual} = Tooling_{Approximate}$

 $Thickness_{OCM}$

7. Repeated this procedure a second or third time for increased accuracy.

For example

You want to deposit 250nm of Al. We have no idea what a good tooling factor is so we use 100 (which is equivalent to uncorrected) as the approximate tooling factor. We deposit what the thickness monitor

tells us is 250nm, but measurement with a profilometer says you actually deposited 157nm. This gives us a tooling factor of 62.8

$$62.8 = 100 \times \frac{157}{250}$$

But we need our deposition to be even more precise therefore repeat the procedure. However we now know that the tooling factor is 62.8 so we use that as our approximate tooling factor and entered it on the thickness monitor. Again we deposit what the thickness monitor says is 250nm. This time the profiler tells us the actual deposition was 263nm. So to calculate an improved tooling factor we enter

$$71.7 = 68.2 \times \frac{263}{250}$$

This gives us a tooling factor of 71.7. You can continue to improve the tooling factor by repeating the process until a satisfactory accuracy is achieved.

When should I check the tooling factor?

- 1. When accuracy is critical.
- 2. When the target (source) material, substrate (sample), or detector placement has changed.
- 3. Large changes to chamber pressure, power, or material density from previous tooling correction.
 - a. I thought the purpose of the tooling factor is a correction for placement of the sensor vs. sample?
 - It is. Unfortunately other parameters can affect the deposition in other ways. For example while a plasma magnetron sputter system is considered fairly isotropic in nature. This often confuses users in to thinking the deposition thickness does not vary spatially with distance from the center. However the plasma is actually focused. Changes to power, pressure and material change the path of source atoms / molecules in their flight to the substrate.
 - 2. If a deposition source did act like a perfect point source, the entire sample stage would have to be hemispherically shaped to keep the distance from the source constant.

Appendix C: Pump / Vent sequence

There is danger of serious damage if pumping and venting of the system is done out of sequence. The good news it's very easy. High vacuum pumps can only work when there is a low vacuum pump on. The basic pump and vent schemes are below (RF cables, process gasses, etc. not included)

Pump down Sequence:

- 1. Turn the Mechanical pump (low vac) on.
- 2. Wait for any reading on the vacuum gauge. A blinking "0000" means vacuum is to poor to read. It won't read until vacuum is 1000mtorrr or lower.
- 3. Turn on the Turbo (high vac) pump. Wait for a stable reading and proceed.

Vent sequence:

- 1. Turn off Turbo pump (high vac).
- 2. Slowly open the vent valve. You will hear a whining sound like a jet engine being turned off. Wait until you don't hear it any more.
- 3. Turn off the Mechanical pump (low vac)
- 4. Once you can open the lid, close the vent valve.